



Biomass

Introduction

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Biomass is an organic, non-fossil material. In other words, biomass describes the mass of all biological organisms, dead or alive, excluding biological mass that has been transformed by geological processes into substances such as coal or petroleum. The entire earth contains about 75 billion tons of biomass. Like coal or petroleum, biomass is a form of stored solar energy. The energy of the sun is captured through the presence of photosynthesis in growing plants. Biomass is sometimes burned as fuel for cooking to produce electricity and heat. This is called biofuel. Like all methods used to generate energy, the combustion of biomass generates polluting as a by-product. Paradoxically, in some industrialized countries like Germany, food is cheaper than fuel compared to the price per joule. Central heating units supplied by food grade wheat or maize are available

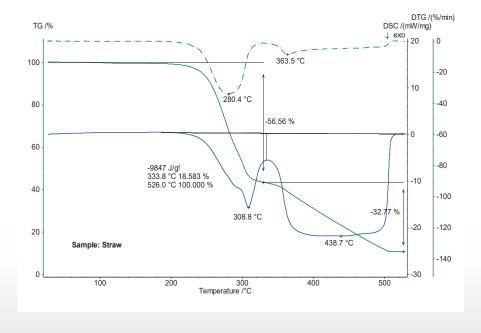
Test Conditions

Temperature range: RT ... 500°C Heating/cooling rates: 20 K/min

Atmosphere: $Ar/20\%/O_2$ at 75 ml/min

Sample mass: 17.26 mg Crucible: Pt

Sensor: $TGA-DSC c_p type S$

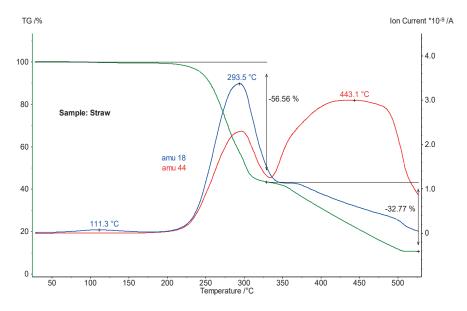


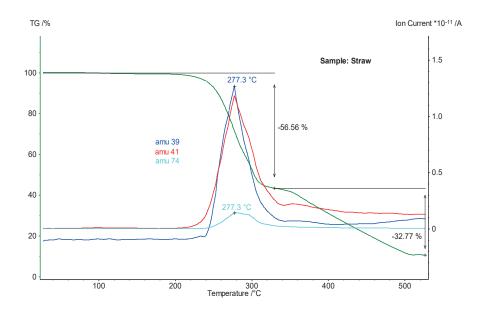


APPLICATION SHEET

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Test Results

The burn-up of straw was studied using simultaneous TGA, DSC and MS. During the burn-up which occurred in two steps of 56.6% and 32.8%, an entire energy of 9.85 kJ/g was released as can be seen from the DSC signal.

The mass-loss steps are mostly due to the release of $\rm CO_2$ (mass number 44) and $\rm H_2O$ (mass number 18) but also salts like KCI (mass number 74) could be detected (see figures below). CO, NO, KOH and $\rm Cl_2$ were furthermore observed which is not shown here for clarity reasons. The residual mass of 10.6% reflects the amount of ash.

